## The Base-catalysed Isomerisation of 2-Acetylimino-3-phenacylthiazolidine

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During synthetic studies on 2-aminothiazoline derivatives, a ready rearrangement of 2-acetylimino-3-phenacylthiazolidine (I)1,2 into the isomeric 2benzoylimino-3-(propan-2-onyl)thiazolidine was observed in a variety of basic media. almost quantitative conversion takes place merely on refluxing with sodium acetate in ethanol. The isomeric compound (II) (m.p. 92°, ex aqueous ethanol) exhibited a strong carbonyl absorption at 1720 cm.-1 and weaker bands at 1625 cm.-1 and 1615 cm.-1 in its infrared spectrum (Nujol). The 60 Mc./sec. n.m.r. spectrum (CDCl<sub>3</sub> solution) indicated the presence of five aromatic protons (multiplets at  $\tau$  1.75 to 1.9 and  $\tau$  2.45 to 2.8), the -CH<sub>2</sub>-CH<sub>2</sub>-system (A<sub>2</sub>X<sub>2</sub> triplets at  $\tau$  6·1 to 6·4 and  $\tau$  6.6 to 6.9), a methylene group (singlet  $\tau$  5.5) and a methyl group (singlet  $\tau$  7.8). Reduction of (II) with sodium borohydride in ethanol furnished the alcohol (III) (m.p. 91°-92° ex toluene), which showed bands at 3510 cm.-1, 1610 cm.-1 and 1600 cm.-1 in its infrared spectrum (Nujol). The 100 Mc./sec. n.m.r. spectrum (CDCl<sub>3</sub> solution shaken with D<sub>2</sub>O) revealed the presence of five aromatic protons (multiplets at  $\tau$  1.75 to 1.95 and  $\tau$  2.5 to 2.8),  $a \ge CH$  group (multiplet at  $\tau$  5.7 to 6.0), a methylene group (part of multiplet at  $\tau$  6.2 to 6.5), the -CH<sub>2</sub>-CH<sub>2</sub>- system [triplets at  $\tau$  6.2 to 6.5 (part of multiplet) and  $\tau$  6.9 to 7.15] and a methyl group (doublet at  $\tau$  8.8 and 8.85). Spin decoupling of the methyl group caused the > CH multiplet to collapse to two doublets representing the X part of an ABX system.

Synthetic verification for the structure of the isomerisation product was provided by condensation of 2-aminothiazoline with chloroacetone to give 2-imino-3-(propan-2-onyl)thiazolidine hydrochloride, which when treated with benzoyl chloride in pyridine furnished a compound identical in every respect with the rearrangement product.

A preliminary investigation of the reaction path suggests the involvement of the diketone (IV), formed via the carbanion generated from (I), as a transient intermediate. Treatment of 1-bromobenzoylacetone3 with two equivalents of 2-aminothiazoline furnished almost one equivalent of 2-aminothiazoline hydrobromide and a brown oil, which on chromatography (1 mm. silica plates, eluent ether) produced (I) and (II) in 13% and 50% yields respectively. All attempts to isolated the hydrobromide of (IV) failed. Extensions of this base-catalysed isomerisation to other systems and its utilisation in heterocyclic syntheses are in progress.

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A. H. M. Raeymaekers, et al., J. Med. Pharm. Chem., 1966, 9, 545.
G. Yoyng and S. I. Crookes, J. Chem. Soc., 1906, 59.
F. Krohnke and H. Timmler, Ber., 1936, 69, 614.